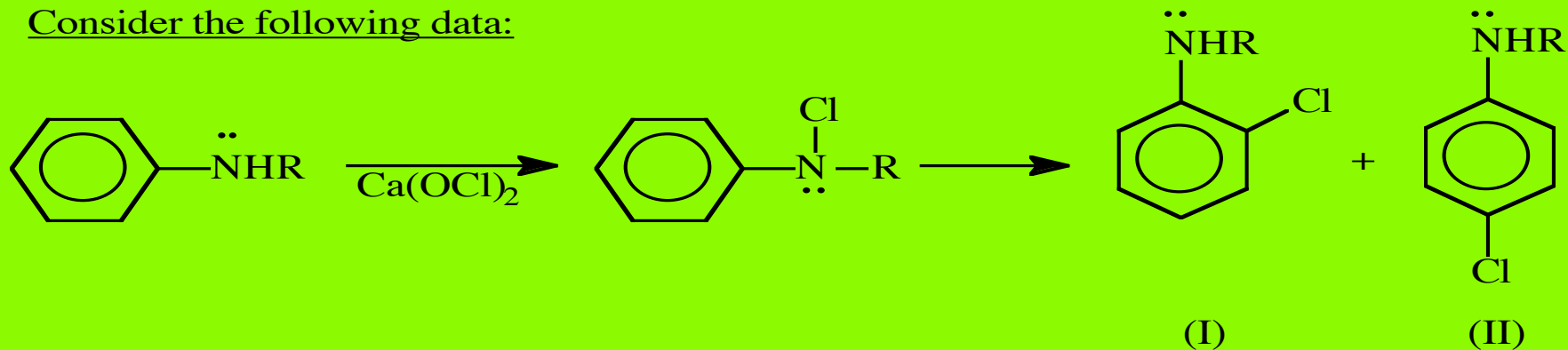
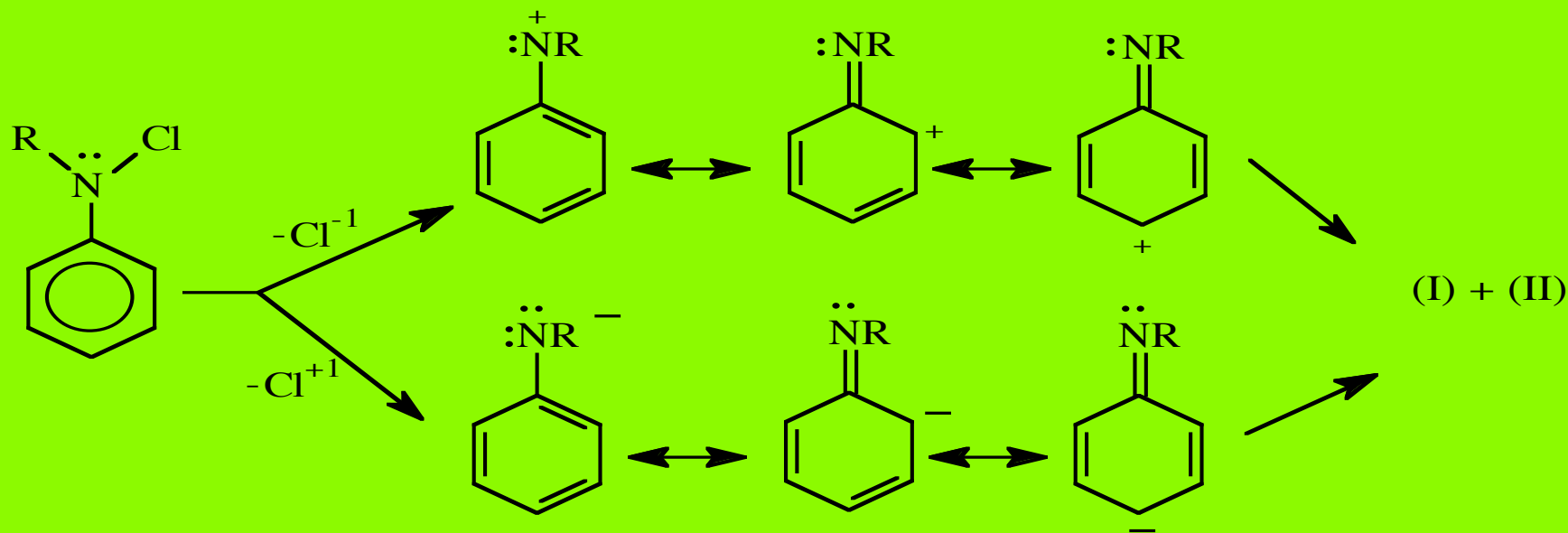


# Carbocation Analogues: Nitrenium ions?

Consider the following data:

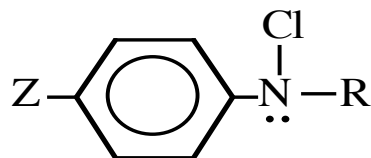


Possible Mechanistic Pathways:



# *Evidence to support the existence of nitrenium ions*

**Study of rearrangement run with para-substituted N-chloroanilines:**

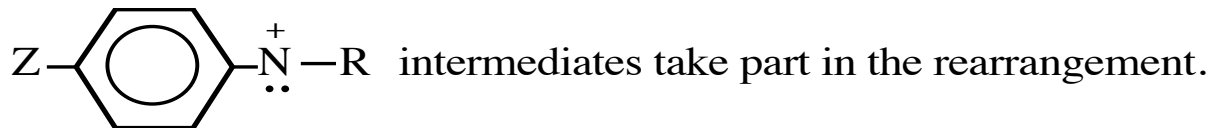


where Z = CH<sub>3</sub>, F, H, Cl, and CN

Observation:

The rearrangement is associated with a **negative rho** value indicative of a reaction favored by electron donating benzene ring substituents.

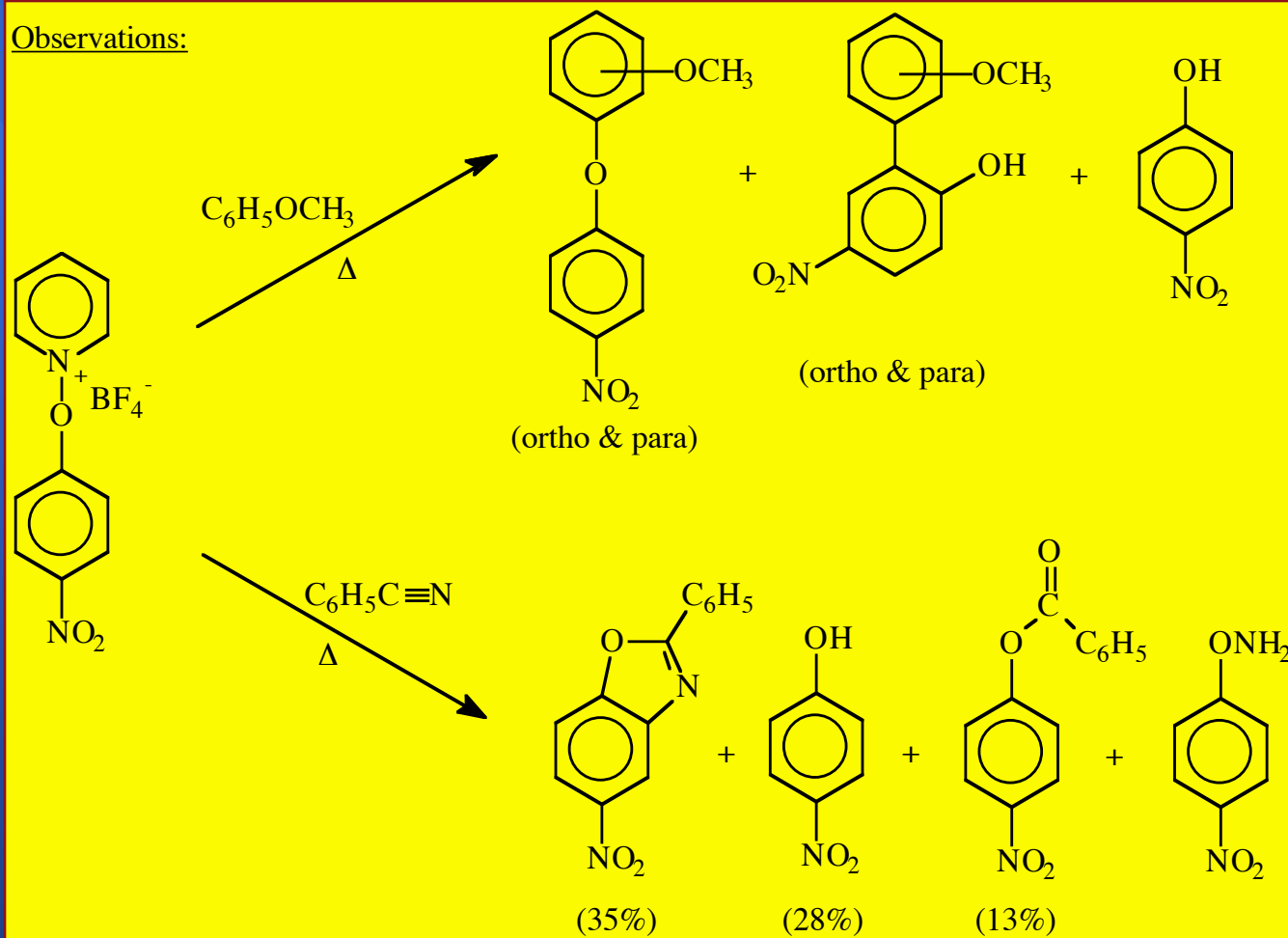
Conclusion:



# Carbocation Analogues

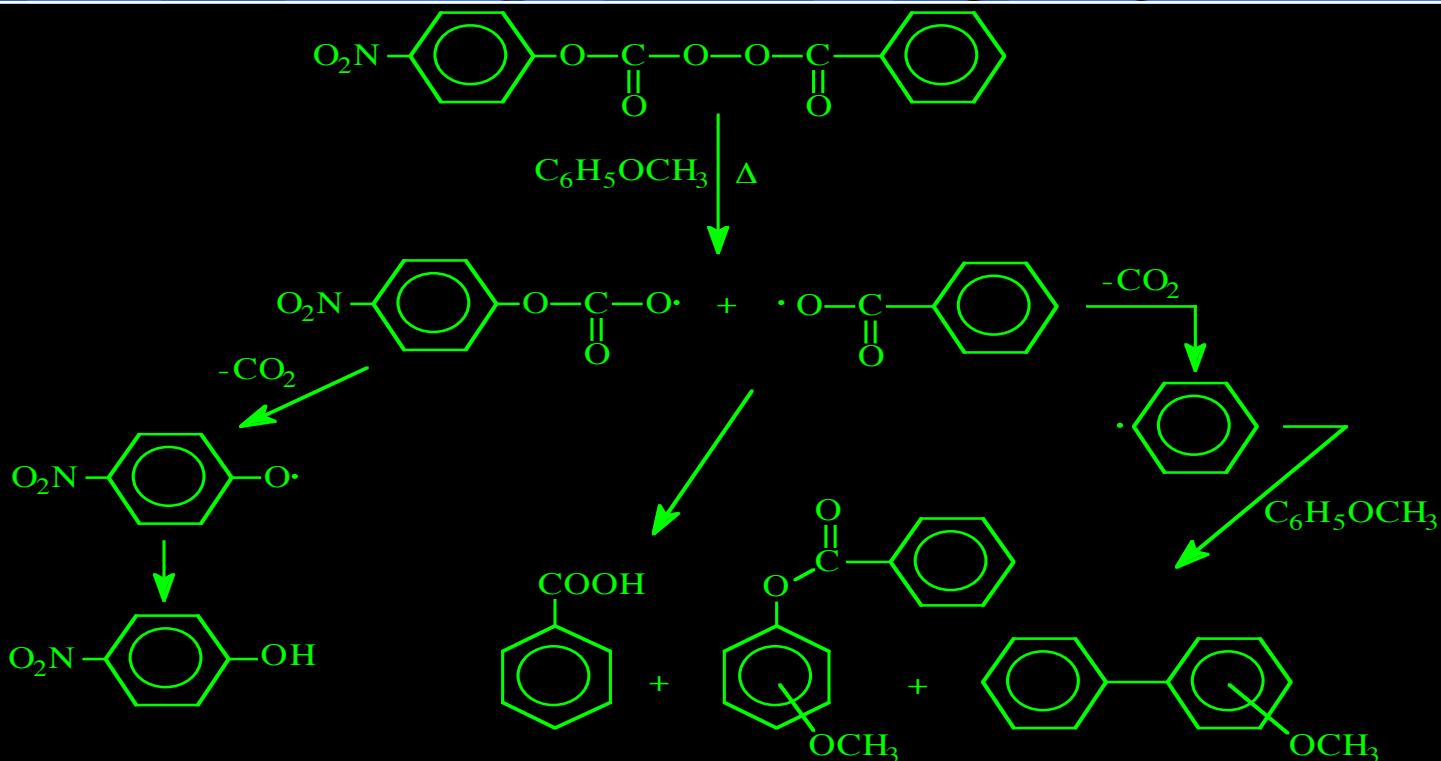
## Oxenium ions?

Observations:

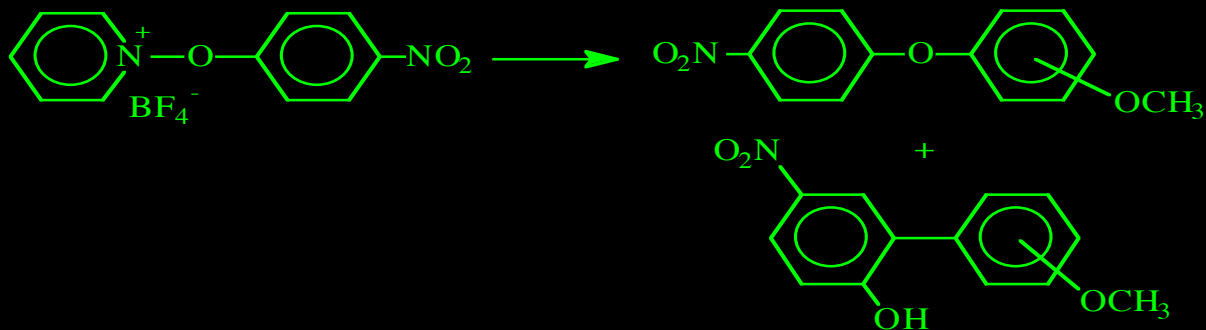


# Relevant Facts (I)

## Evidence against a free radical pathway



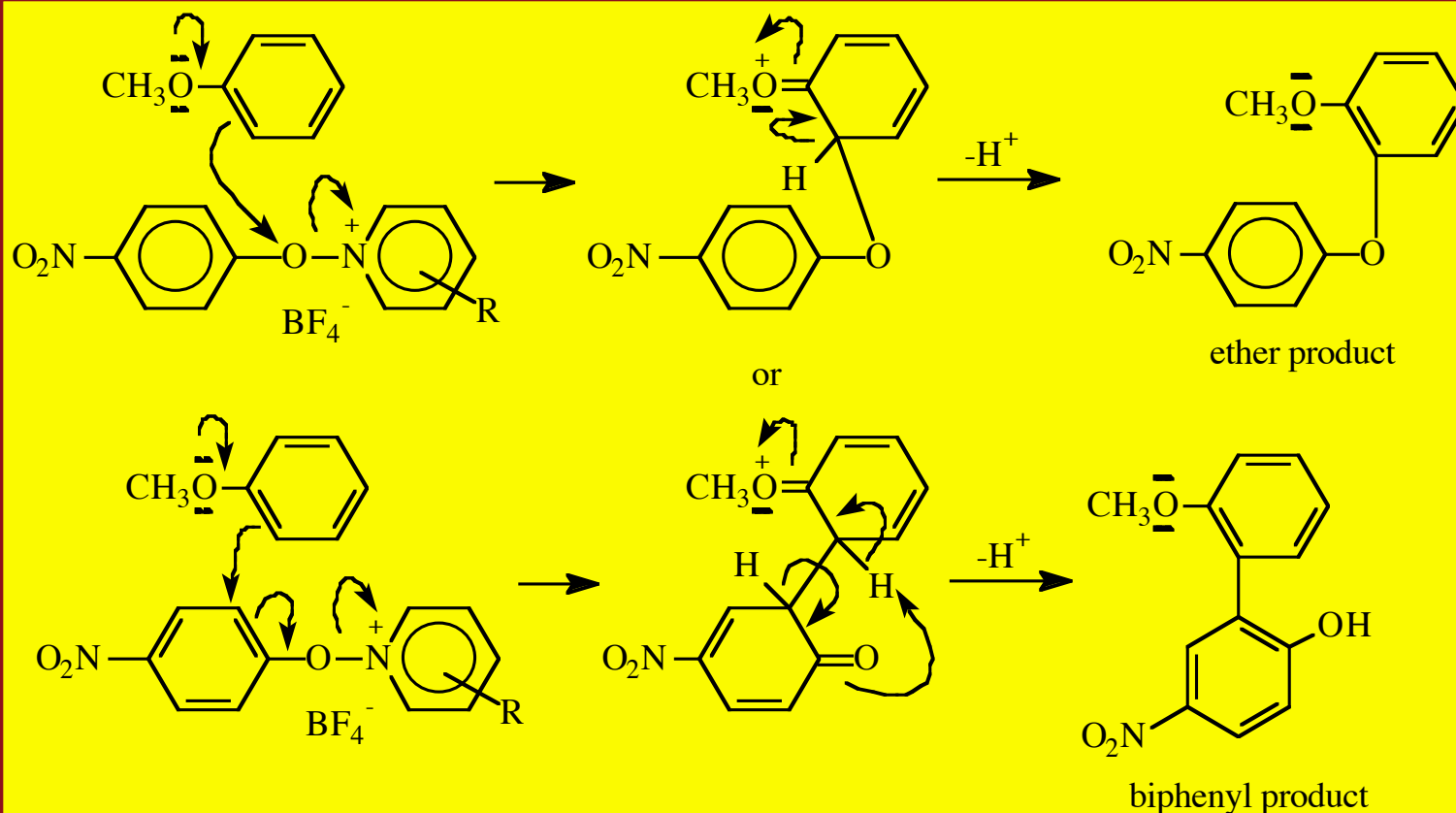
Under the above conditions:



**These products are not produced from radical intermediates under identical reaction conditions.**

# Relevant Facts (II)

## Evidence against a concerted pathway

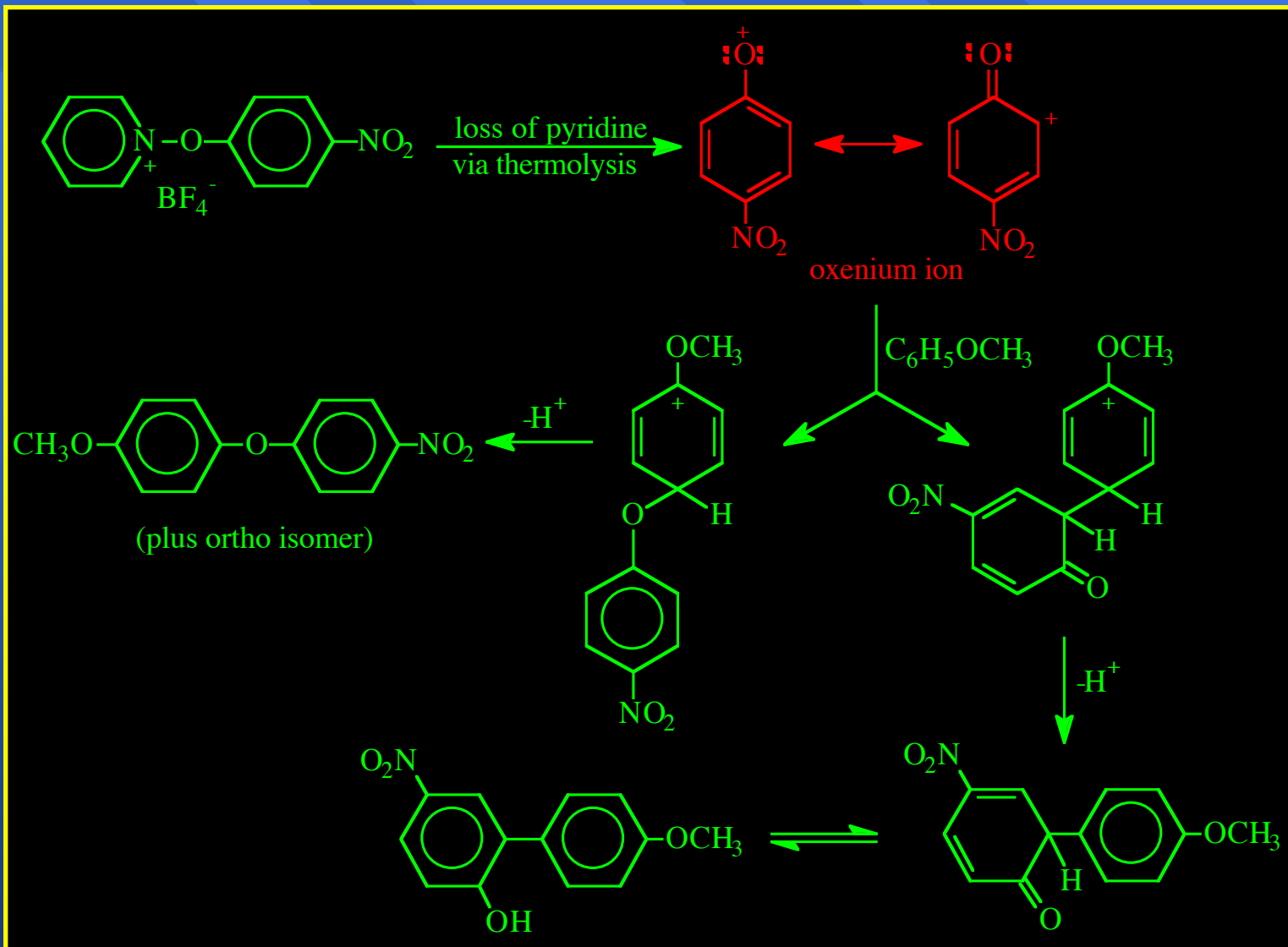


In a concerted pathway, the pyridine ring substituent would be expected to affect the leaving ability of the pyridine group and thereby have an impact on the ratio of ether:biphenyl products.

Observation: For R = 4-methyl, 4-methoxy, 4-phenyl, and 2-hydroxy, a constant ratio of ortho to para diphenyl ethers (28%:72%) was obtained. A constant ratio of ortho to para biphenyl products (75%:25%) was obtained.

Conclusion : The evidence suggests that attack is independent of the pyridine ring R-substituent, a fact contrary to what one would expect for a concerted mechanism.

# A plausible explanation involving the intermediacy of Oxenium ions



# A plausible explanation involving the intermediacy of Oxenium ions cont' d.

