NT		
Name		

(36) 1. Supply the missing reagents in the following transformations:

a) ethyne
$$\frac{}{?}$$
 1-hexyne $\frac{}{?}$ CH₃(CH₂)₃-C-CH₃

b) 2-bromopropane propene 3-bromopropene
?
| ? | ? | ?
| 1-bromopropane 1,2,3-tribromopropane

c)
$$CH \xrightarrow{?} Cyclohexene \xrightarrow{?} CH \text{ (meso)}$$

$$H-C-(CH_2)_4-C-H \text{ II O O O}$$

d) 4-octyne
$$\frac{}{?}$$
 E-4-octene $\frac{}{?}$ 2 CH₃CH₂COOH

(6) 2. The alarm pheromone of the green peach aphid has the formula C₁₅H₂₄. Upon catalytic hydrogenation, C₁₅H₂₄ is converted to 2,6,10-trimethyldodecane. Ozonolysis with reductive workup of C₁₅H₂₄ produces the cleavage products shown below:

Propose a structure for $C_{15}H_{24}$ based on the data provided. Ignore the possibility of E/Z isomerism.

(22) 3. Supply the missing reactants/products in the following reactions:

a) or
$$H_2^+$$
 CH_3 CH_3 CH_3 CH_3 CH_3 CH_3

b)
$$\frac{\text{H}_{3}\text{C} \text{ CH}_{3} \overset{\text{O}}{\text{O}}}{\text{dienophile}} + \frac{\text{diene}}{\text{dienophile}}$$

c) 3,4-dibromohexane
$$\frac{Zn/Alc.}{\Delta}$$

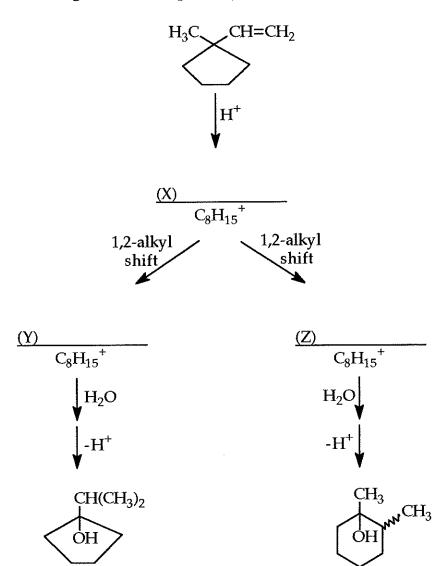
d)
$$alkynyl anion$$
 $CH_3CH_2CH_2-C\equiv C$ $+$ $+$ NaI $substrate$

(12) 4. Achiral compounds (L) and (M) have the molecular formula C₇H₁₄ and are E/Z diasteromers with respect to one another. Catalytic hydrogenation of either (L) or (M) produces a racemic mixture of compound (N) of molecular formula C₇H₁₆.

Hint: Hydrogenation of the (L)/(M) diastereomers leads to the creation of a stereogenic center in compound (N).

Propose plausible structures for (L), (M), and (N). From the data provided, it will not be possible to differentiate between (L) and (M).

(12) 5. Draw the structures of the carbocationic intermediates in the following mechanistic pathway:



a mixture of diastereomers

(12) 6. Consider the dehydrohalogenation reaction shown below:

For the one elimination pathway by which the above reaction proceeds (E-1? or E-2?), will it be true or false that:

- a) (I) represents the major alkene product
- b) alkene formation occurs via a carbocationic intermediate
- c) the rate of elimination is directly proportional to the concentration of alkyl bromide reactant
- d) the rate of elimination will be enhanced if bromine is replaced by iodine