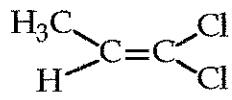
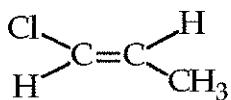


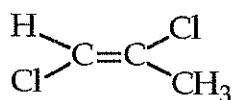
- (20) 1. Consider the compounds shown below:



(I)



(II)



(III)

- a) Which one of the above compounds would exhibit a different number of pmr signals than the other two? _____
- b) Which one of the above compounds would have the most upfield pmr signal? _____
- c) Which one of the above compounds would have the most downfield pmr signal? _____
- d) Which one of the above compounds would have the same number of pmr and cmr signals? _____
- e) Which one of the above compounds would have the most downfield cmr signal? _____

- (12) 2. Draw the structures of three constitutional isomers of formula $\text{C}_2\text{H}_2\text{Br}_2\text{Cl}_2$ whose pmr spectra consist of only a singlet signal.
- _____
- _____
- _____

(15) 3. Draw structural formulas for three of the four compounds shown below based on the pmr spectral data provided.

a)	$C_5H_{10}Br_2$	singlet	δ 1.2 ppm	6H	_____
		triplet	δ 2.0 ppm	2H	
		triplet	δ 3.2 ppm	2H	
b)	$C_7H_{14}O_2$	singlet	δ 1.1 ppm	9H	_____
		singlet	δ 2.1 ppm	3H	
		singlet	δ 4.2 ppm	2H	
c)	$C_5H_9BrO_2$	triplet	δ 1.2 ppm	3H	_____
		triplet	δ 2.9 ppm	2H	
		triplet	δ 3.5 ppm	2H	
		quartet	δ 4.0 ppm	2H	
d)	$C_5H_{10}O_3$	triplet	δ 1.3 ppm	6H	_____
		quartet	δ 4.1 ppm	4H	

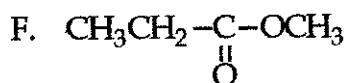
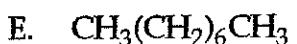
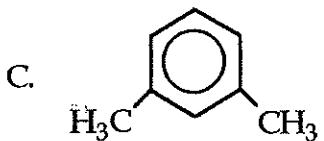
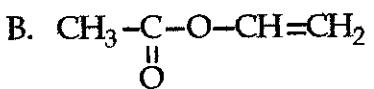
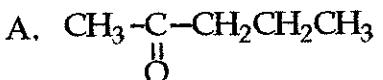
- (12) 4. For the four sets of proton-decoupled ¹³C NMR spectra that follow, select a structure from those shown below that corresponds to the appropriate spectral data set.

Cmr data set #1: δ 22.7 ppm
δ 25.0 ppm
δ 41.8 ppm
δ 60.5 ppm

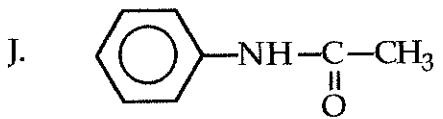
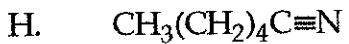
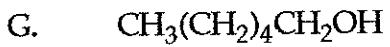
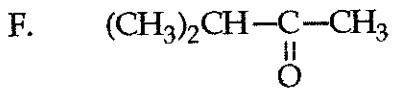
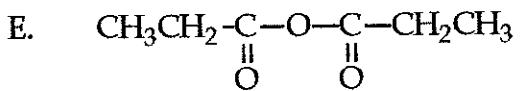
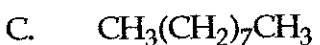
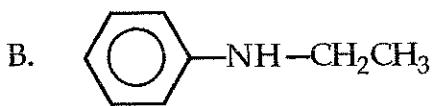
Cmr data set #2: δ 14.1 ppm
δ 22.9 ppm
δ 28.4 ppm
δ 32.7 ppm
δ 62.4 ppm

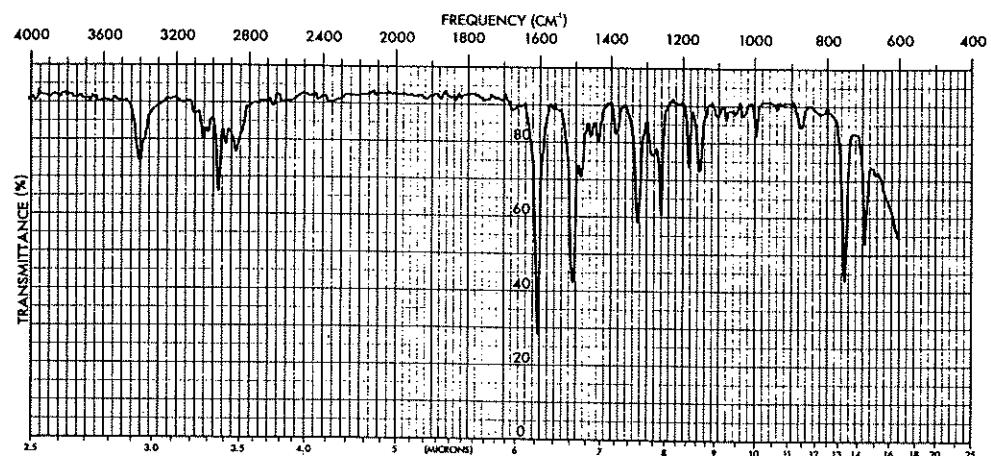
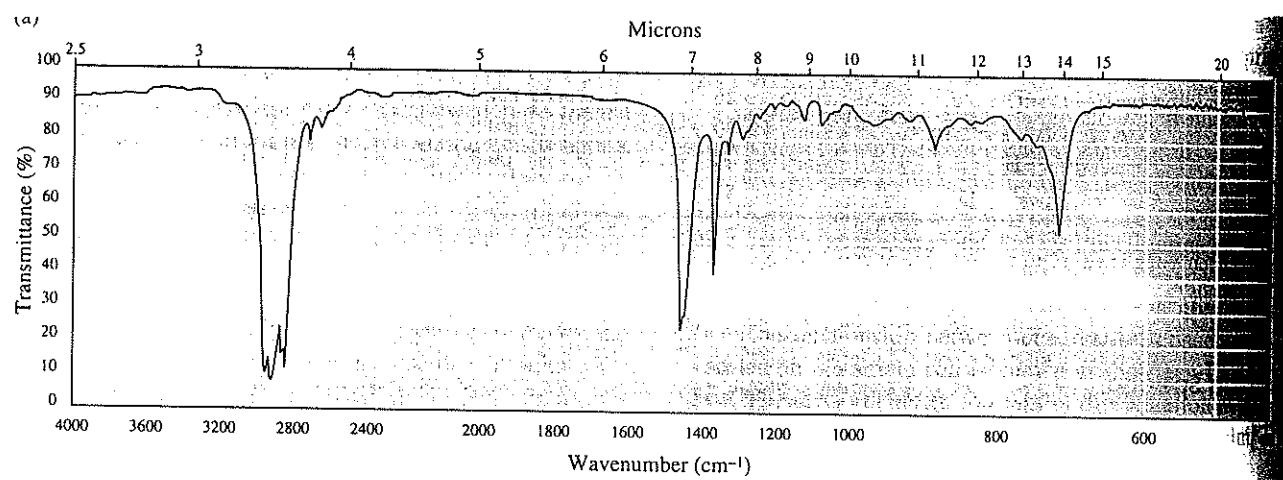
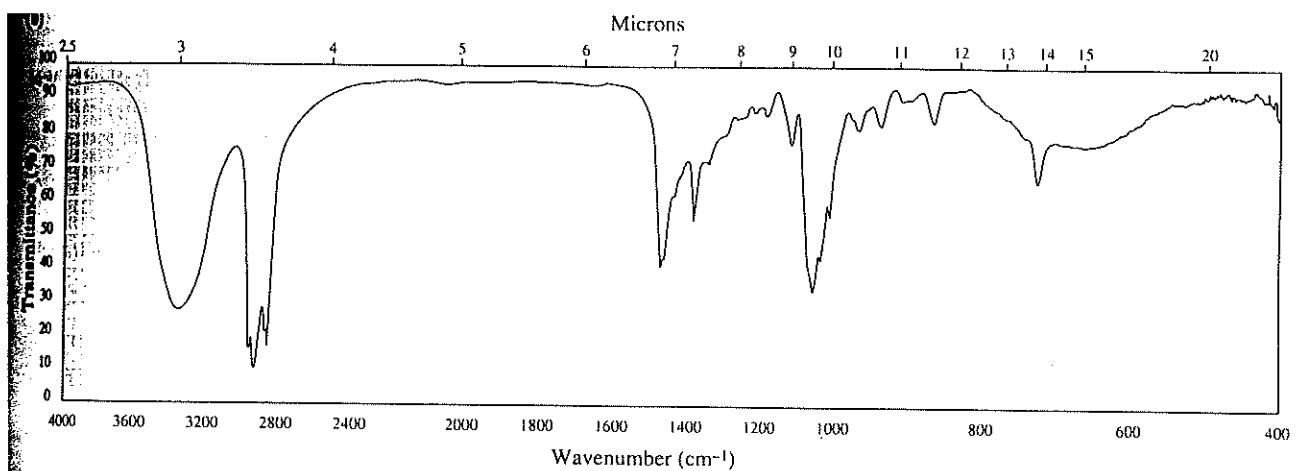
Cmr data set #3: δ 20.4 ppm
δ 98.2 ppm
δ 141.7 ppm
δ 167.7 ppm

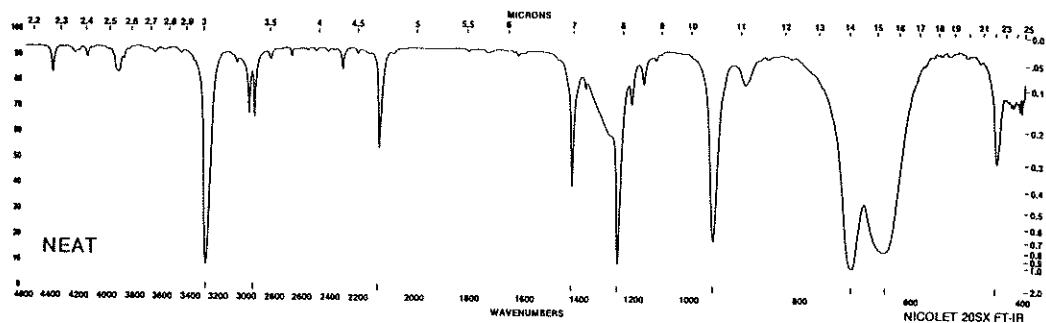
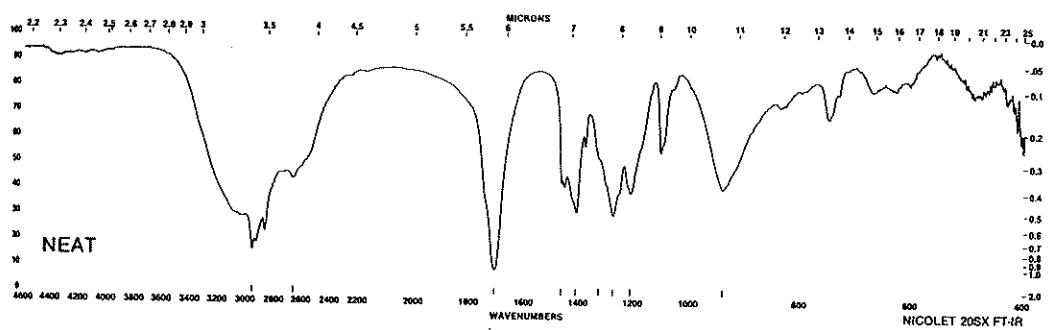
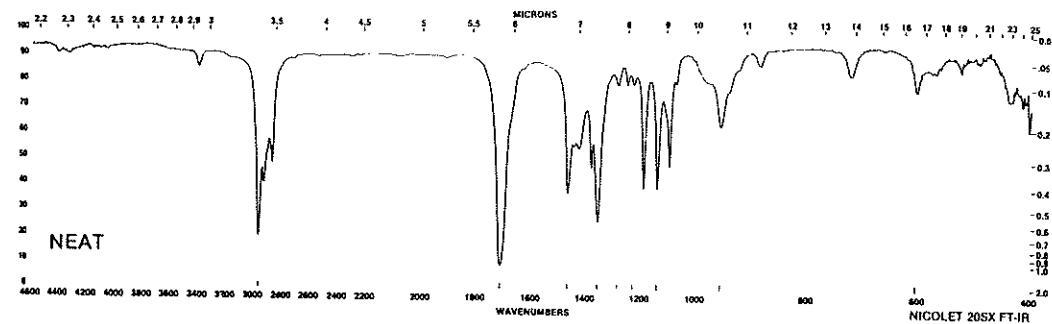
Cmr data set #4: δ 10.3 ppm
δ 27.6 ppm
δ 51.4 ppm
δ 174.6 ppm



(24) 5. For the six infrared spectra that follow, specify which spectra correspond to which of the compounds below:







- (5) 6. An unknown compound of formula C₆H₁₀ reacts with excess H₂ in the presence of platinum to give C₆H₁₂. The proton-coupled cmr spectrum of the unknown consists of the following signals:

triplet	@	δ 22.9 ppm
triplet	@	δ 25.3 ppm
doublet	@	δ 127.2 ppm

Based on the data provided, deduce the structure of the unknown compound.

- (12) 7. Based on the spectral data provided below, deduce the structures of (A) and (B):

(A)



Sig. IR: 1735 cm⁻¹

Pmr: 6H (t) δ 1.20 ppm
4H (s) δ 2.61 ppm
4H (q) δ 4.15 ppm

(B)

Molar mass = 83 g/mol

Sig. IR: 2250 cm⁻¹

Pmr: 6H (d) δ 1.07 ppm
1H (m) δ 2.03 ppm
2H (d) δ 2.26 ppm

(A)

(B)