# The von Richter Reaction: A Case History

In 1871, von Richter reported the following observations:

$$NO_2$$
  $KCN$   $EtOH/H_2O$   $COOH + KNO_2$  reflux

$$Z - \underbrace{\begin{array}{c} KCN \\ EtOH/H_2O \\ reflux \end{array}} - COOH + "acidic tars" + KNO_2$$

5%-50% yields

#### Based on the above:

- 1. Direct displacement of the nitro group is eliminated from consideration.
- 2. A benzyne intermediate is ruled out. In the case of para-substituted nitrobenzenes, benzyne intermediates would give both meta- and para-substituted benzoic acids.

## von Richter Reaction: Bunnett Mechanism (1954)

In 1954, Bunnett et al. proposed the following mechanism to account for the known facts pertaining to the von Richter reaction:

Z-NO<sub>2</sub> + CN Z-NO<sub>2</sub>
H-OH

$$Z$$
-NO<sub>2</sub> + CN Z-NO<sub>2</sub>
H + NO<sub>2</sub>
 $Z$ -NO<sub>2</sub>
 $Z$ -NO<sub>3</sub>
 $Z$ -NO<sub>4</sub>
 $Z$ -NO<sub>2</sub>
 $Z$ -NO<sub>2</sub>
 $Z$ -NO<sub>3</sub>
 $Z$ -NO<sub>4</sub>
 $Z$ -NO<sub>2</sub>
 $Z$ -NO<sub>4</sub>
 $Z$ -NO<sub>5</sub>
 $Z$ -NO<sub>6</sub>
 $Z$ -NO<sub>7</sub>
 $Z$ -N

#### von Richter Reaction: Bunnett Observations (1956)

### VON RICHTER REACTION: REVISED BUNNETT MECHANISM

#### Rosenblum Observations (1960)

Observation I: Instead of nitrite, molecular nitrogen was a by-product of the von Richter reaction. Apparently, in the 1871 von Richter paper, nitrite had never been demonstrated to be a by-product. It had been deduced based on stoichiometry considerations.

In the revised Bunnett mechanism, ammonia and nitrite are stipulated by-products of the von Richter reaction. Ammonia and nitrite can react to form ammonium nitrite which, upon heating, decomposes to give molecular nitrogen and water.

Upon addition of  ${}^{15}NH_3$  to an on-going von Richter reaction, the only molecular nitrogen obtained contained no nitrogen-15! If the revised Bunnett mechanism were correct, the added  ${}^{15}NH_3$  and the expelled  ${}^{16}N = {}^{14}N$ .

Observation II:

Point of Interest:

However:

CI 
$$\longrightarrow$$
 NO<sub>2</sub>  $\xrightarrow{\text{KCN}}$  CI  $\longrightarrow$  CI  $\longrightarrow$  +  $^{15}\text{N} \stackrel{\text{14}}{=} \text{N}$  reflux

Conclusion: One nitrogen atom must come from the nitro group, and the second must come from the cyano group **intramolecularly**!

### Accepted Mechanism for the von Richter Reaction

## Supporting Evidence for the Rosenblum Mechanism

#### Ullman & Bartkus (1962): NH Pb(OAc)<sub>4</sub> + $N_2$ EtOH/H<sub>2</sub>O reflux hydrazide diazene Ibne-Rasa & Koubek (1963): $NH_2$ RCO<sub>3</sub>H $+ N_2$ EtOH/H2O reflux $-NH_2$ C-NH<sub>2</sub> o-nitrosobenzamide